

# Blood Gases

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# Interpretation of blood gases

- ▶ Acid base balance
  - ▶ pH, CO<sub>2</sub>, HCO<sub>3</sub>
  - ▶ Major disturbance
  - ▶ Compensation
  - ▶ Metabolic acidosis
    - ▶ Anion gap
    - ▶ Delta ratio
- ▶ Oxygenation
- ▶ Other electrolytes

# Arterial Blood Gas Analysis

ABG Parameter			ABG result	Calculation and interpretation		
<b>pH</b>	>7.45	Alkalaemia		<b>pH</b>	<b>pCO2</b>	<b>Interpretation</b>
	<b>7.36-44</b>	Normal		↓	↓	Metabolic acidosis
	<7.35	Acidaemia		↑	↑	Metabolic alkalosis
<b>pCO2</b>	>45	High		↑	↓	Respiratory alkalosis
	<b>35-45</b>	Normal		↓	↑	Respiratory acidosis
	<35	Low		<b>Corrected standard AG for albumin</b>		
<b>HCO3</b>	>26	High		$\frac{\text{Albumin} + 1.5 \times \text{Phosphate}}{4}$		
	<b>24+/- 2</b>	Normal		<b>Anion Gap calculation</b>		
	<22	Low		$\{[\text{Na}^+] - [\text{Cl}^- + \text{HCO}_3]\} = 12 \pm 4$		
<b>AG</b>	> 16	High		<b>Corrected Na+ for AG in hyperglycemia</b>		
	<b>12+/-4</b>	Normal		$\text{Corrected Na}^+ = \text{Na} + \frac{\text{Glucose} - 5}{3}$		
	< 8	Low		<b>Gap: Gap calculation for metabolic acidosis</b>		
<b>Glucose</b>	>10	High		<0.4	Low or Normal AG metabolic acidosis	
	< 2	Low		0.4-0.8	Normal + high AG metabolic acidosis	
<b>Gap: Gap</b>	$\frac{\Delta \text{AG}}{\Delta \text{HCO}_3} = \frac{\text{AG} - 12}{24 - \text{HCO}_3}$			0.8-2.0	Pure high metabolic acidosis	
				>2.0	Metabolic acidosis with metabolic alkalosis/respiratory acidosis	
				<b>PAO2 = [713 x FiO2] - [pCO2 x 1.25]</b>		
<b>Lactate</b>	<1.9	Normal		<b>A-a gradient = PAO2 - PaO2 = <math>\frac{\text{Age} + 4}{4}</math></b>		
	>2.0	High				
<b>pO2</b>	80-100	Normal				
	< 80	Hypoxia				

### Compensation rules for

Expected PCO2	Metabolic acidosis		Metabolic alkalosis	
		$1.5 \times [\text{HCO}_3] + 8 \text{ (+/- 2)}$		$0.7 \times [\text{HCO}_3] + 20 \text{ (+/- 5)}$
Expected HCO3	Respiratory acidosis		Respiratory alkalosis	
	Acute	Chronic	Acute	Chronic
	$24 + \frac{\text{pCO}_2 - 40}{10} \times 1$	$24 + \frac{\text{pCO}_2 - 40}{10} \times 4$	$24 - \frac{40 - \text{pCO}_2}{10} \times 2$	$24 - \frac{40 - \text{pCO}_2}{10} \times 5$

# Acid Base Balance

## ▶ pH

- ▶ Normal 7.35-7.45
- ▶ Alkalaemia >7.45
- ▶ Acidaemia <7.35

## ▶ pCO<sub>2</sub>

- ▶ Normal 35-45
- ▶ High >45
- ▶ Low <35

## ▶ HCO<sub>3</sub>

- ▶ Normal 24 +/- 2
- ▶ High >26
- ▶ Low <22

# Major Disturbance



**ROME**

# Major Disturbance

	pH	CO <sub>2</sub>	HCO <sub>3</sub>
Metabolic acidosis	↓		↓
Respiratory acidosis	↓	↑	
Metabolic alkalosis	↑		↑
Respiratory alkalosis	↑	↓	



- ▶ 63f with dyspnoea

- ▶ pH 7.29

- ▶ PCO<sub>2</sub> 68

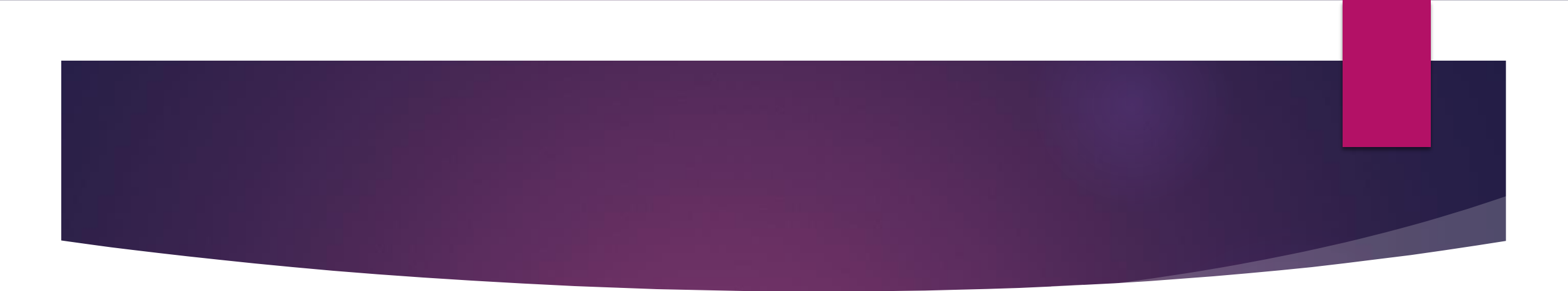
- ▶ HCO<sub>3</sub> 26

- ▶ Acidaemia

- ▶ Hypercarbia

- ▶ Normal HCO<sub>3</sub>

- ▶ **Respiratory acidosis**



- ▶ 17f chest tightness, dyspnoea, perioral tingling, upper limb paraesthesias

- ▶ pH 7.49

- ▶ PCO<sub>2</sub> 24

- ▶ HCO<sub>3</sub> 22

- ▶ Alkalaemia

- ▶ Hypocarbia

- ▶ Normal HCO<sub>3</sub>

- ▶ **Respiratory alkalosis**





- ▶ 48m distended abdo + vomiting. CT shows bowel obstruction

- ▶ pH 7.5

- ▶ PCO<sub>2</sub> 41

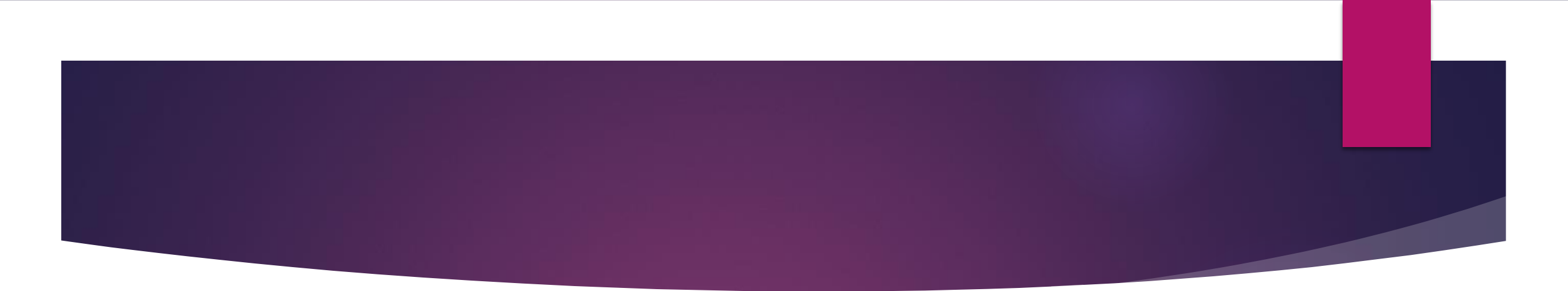
- ▶ HCO<sub>3</sub> 29

- ▶ Alkalaemia

- ▶ Normal CO<sub>2</sub>

- ▶ High HCO<sub>3</sub>

- ▶ **Metabolic alkalosis**



- ▶ 89f confusion, fevers, rigors, hypotension, oliguria

- ▶ pH 7.29

- ▶ PCO<sub>2</sub> 41

- ▶ HCO<sub>3</sub> 15

- ▶ Acidaemia

- ▶ Normal CO<sub>2</sub>

- ▶ Low HCO<sub>3</sub>

- ▶ **Metabolic acidosis**

# Compensation

## Expected PCO<sub>2</sub>

- ▶ Metabolic acidosis
  - ▶  $1.5 \times \text{HCO}_3 + 8 (+/-2)$
  - ▶ *Winter's equation*
- ▶ Metabolic alkalosis
  - ▶  $0.7 \times \text{HCO}_3 + 20 (+/-5)$

## Expected HCO<sub>3</sub>

	HCO <sub>3</sub> (baseline 24mmol/L)	
Every <b>10mmHg</b> change in PCO <sub>2</sub> from baseline 40mmHg	Acute	Chronic
Increased PCO <sub>2</sub> (resp acidosis)	<b>1</b>	<b>4</b>
Decreased PCO <sub>2</sub> (resp alkalosis)	<b>2</b>	<b>5</b>

- ▶ 56m found by bystanders with GCS7 and pinprick pupils. Naloxone administered by paramedics

- ▶ pH 7.31
- ▶ PCO<sub>2</sub> 55
- ▶ HCO<sub>3</sub> 27

- ▶ Acidaemia
- ▶ Hypercarbia
- ▶ High HCO<sub>3</sub>      Major disturbance?
- ▶ Expected HCO<sub>3</sub>
  - ▶ Acute 26
  - ▶ Chronic 30

- ▶ **Acute respiratory acidosis with metabolic compensation**

▶ 77f 10/7 post #NOF with hypoxia and tachypnoea

▶ pH 7.51

▶ PCO<sub>2</sub> 23

▶ HCO<sub>3</sub> 21

▶ Alkalaemia

▶ Hypocarbia

▶ Low HCO<sub>3</sub>                      Major disturbance?

▶ Expected HCO<sub>3</sub>

▶ Acute 20

▶ Chronic 14

▶ **Acute respiratory alkalosis with metabolic compensation**

▶ 24m medical student returned from elective in Nepal. Few days severe diarrhoea with dehydration and tachypnoea

▶ pH 7.32

▶ PCO<sub>2</sub> 30

▶ HCO<sub>3</sub> 13

▶ Acidaemia

▶ Hypocarbia

▶ Low HCO<sub>3</sub>                      Major disturbance?

▶ Expected CO<sub>2</sub>

▶  $1.5 \times 13 + 8 = 28$

▶ **Metabolic acidosis with respiratory compensation**

▶ 34f presents with vomiting after binge drinking last night

▶ pH 7.46

▶ PCO<sub>2</sub> 52

▶ HCO<sub>3</sub> 39

▶ Alkalaemia

▶ Hypercarbia

▶ High HCO<sub>3</sub>      Major disturbance?

▶ Expected CO<sub>2</sub>

▶  $0.7 \times \text{HCO}_3 + 20 (+/-5) = 47 (+/-5)$

▶ **Metabolic alkalosis with respiratory compensation**

▶ 27m footballer completing pre-season training in Colorado at high altitude

▶ pH 7.47

▶ PCO<sub>2</sub> 20

▶ HCO<sub>3</sub> 14

▶ Alkalaemia

▶ Hypocarbia

▶ Low HCO<sub>3</sub>           Major disturbance?

▶ Expected HCO<sub>3</sub>

▶ Acute 22

▶ Chronic 14

▶ **Chronic respiratory alkalosis with metabolic compensation**



- ▶ 64 y/o male
- ▶ End stage COPD
- ▶ Currently smoking
- ▶ 3/7 worsening dyspnoea

**BLOOD GAS ANALYSIS**

**SPECIMEN: BLOOD**

Date: 19/08/20 19/08/20 19/08/20  
 Coll. Time: 17:04 16:02 14:38  
 Lab Number: 43472555 43472552 43472648 Ref. Range Unit

Blood Type	Venous	Venous	Venous		
Temperature	37.5	37.0	37.0		
pH	7.22	7.24	7.31	(7.35 - 7.45)	
pCO2	111	104	84	(35 - 46)	mm Hg
pO2	39	38	27	(80 - 100)	mm Hg
T corr pH	7.21	7.24	7.31	(7.35 - 7.45)	
T corr pCO2	113	104	84	(35 - 46)	mm Hg
T corr pO2	40	38	27	(80 - 100)	mm Hg
HC03	45	45	42	(21 - 28)	mmol/l
Base Ex	12.2	12.0	11.5	(-2 - +3)	mmol/l
sO2	66.2	61.3	44.9	(95.0 - 100.0)%	
Sodium	139	138	135	(135 - 145)	mmol/L
Potassium	4.4	4.3	4.2	(3.4 - 4.8)	mmol/L
Chloride	97	96	96	(95 - 110)	mmol/L
Anion Gap	1	2	1	(8 - 20)	mmol/L
Ionised Ca	1.18	1.17	1.11	(1.15 - 1.35)	mmol/L
Glucose	6.9	8.3	10.2	(3.0 - 6.9)	mmol/L
L-Lactate	1.1	1.3	1.7	(< 1.5)	mmol/L
Haemoglobin	148	150	157	(120 - 180)	g/L
Haematocrit	44	45	47	(35 - 51)	%
O2Hb	63.1	58.4	43.1	(> 95.0)	%
COHb	3.4	3.0	2.9	(< 2.0)	%
MetHb	1.2	1.7	1.2	(< 3.0)	%
HHb	32.2	36.9	52.9	(< 5.0)	%
FIO2	25				%

- ▶ Acidaemia
- ▶ Hypercarbia
- ▶ High HCO3 Major disturbance?
- ▶ Expected HCO3
  - ▶ Acute 28
  - ▶ Chronic 40
- ▶ **Chronic respiratory acidosis with metabolic compensation**
- ▶ Following deterioration?
- ▶ **Acute on chronic respiratory acidosis with metabolic compensation**

▶ 64m chest pain, arrests soon after arrival receiving 6min CPR and regaining ROSC

▶ pH 7.14

▶ PCO<sub>2</sub> 60

▶ HCO<sub>3</sub> 15

▶ Acidaemia

▶ Hypercarbia

▶ Low HCO<sub>3</sub>      Major disturbance?

▶ **Respiratory acidosis and metabolic acidosis**

# Metabolic acidosis – Anion gap

▶  $(\text{Na}^+ + \text{K}^+) + (\text{Cl}^- + \text{HCO}_3^-)$

▶ Anion Gap

▶ Normal 12 +/- 4

▶ High >16

▶ Low <8

# High Anion Gap Metabolic Acidosis – Delta Ratio

$$\blacktriangleright \frac{\Delta AG}{\Delta HCO_3} = \frac{AG - 12}{24 - HCO_3}$$

Delta Ratio	
<0.4	NAGMA
0.4-0.8	NAGMA + HAGMA
0.8-2	HAGMA
>2	Metabolic acidosis + metabolic alkalosis

**BLOOD GAS ANALYSIS**

**SPECIMEN: BLOOD**

Date: 19/08/20 19/08/20 19/08/20  
 Coll. Time: 18:07 16:43 15:07  
 Lab Number: 43472559 43472554 43472649 Ref. Range Unit

Blood Type	Venous	Venous	Venous		
Temperature	37.0	37.0	37.0		
pH	7.32	7.25	7.18	(7.35 - 7.45)	
pCO2	31	29	31	(35 - 46)	mm Hg
pO2	70	63	28	(80 - 100)	mm Hg
T corr pH	7.32	7.25	7.18	(7.35 - 7.45)	
T corr pCO2	31	29	31	(35 - 46)	mm Hg
T corr pO2	70	63	28	(80 - 100)	mm Hg
HC03	16	13	12	(21 - 28)	mmol/l
Base Ex	-8.9	-13.1	-15.5	(-2 - +3)	mmol/l
sO2	95.7	93.5	61.2	(95.0 - 100.0)%	
Sodium	133	131	129	(135 - 145)	mmol/L
Potassium	3.8	3.8	4.6	(3.4 - 4.8)	mmol/L
Chloride	106	104	98	(95 - 110)	mmol/L
Anion Gap	15	18	24	(8 - 20)	mmol/L
Ionised Ca	1.09	1.09	1.11	(1.15 - 1.35)	mmol/L
Glucose	5.2	7.1	18.0	(3.0 - 6.9)	mmol/L
L-Lactate	0.8	1.3	1.8	(< 1.5)	mmol/L
Haemoglobin	140	141	152	(120 - 180)	g/L
Haematocrit	42	42	46	(35 - 51)	%
O2Hb	91.8	89.8	58.9	(> 95.0)	%
COHb	2.6	2.7	2.2	(< 2.0)	%
MetHb	1.5	1.3	1.5	(< 3.0)	%
HHb	4.1	6.2	37.4	(< 5.0)	%

▶ 23m T1DM presents following fist fight with his brother. Missed insulin dose that morning due to stress

- ▶ Acidaemia
- ▶ Hypocarbia
- ▶ Low HCO3 Major disturbance?
- ▶ Expected CO2: 26
- ▶  $AG = (Na^+ + K^+) - (Cl^- + HCO3^-) = 23$
- ▶  $\Delta \text{ ratio} = \frac{AG - 12}{24 - HCO3} = 0.9$
- ▶ **HAGMA with incomplete respiratory compensation**

# Causes of HAGMA

- ▶ Lactate
  - ▶ Hypoperfusion
  - ▶ Medical problems
    - ▶ Sepsis
    - ▶ Liver failure
  - ▶ Drugs
    - ▶ Metformin
    - ▶ iron
  - ▶ Inborn errors of metabolism
- ▶ Toxins
  - ▶ Salicylates
- ▶ Ketones
  - ▶ DKA
  - ▶ ETOH
  - ▶ Starvation
- ▶ Renal failure
  - ▶ Urea



**Left Total Knee Replacement**

# Causes of NAGMA



**HARDUP**

- ▶ **Hyperchloraemia**
- ▶ **Addison's disease**
- ▶ **Drugs**
  - ▶ Carbonic anhydrase inhibitors eg. **Acetazolamide**
  - ▶ Spironolcatone
- ▶ **HCO<sub>3</sub> loss**
  - ▶ **Renal Tubular Acidosis**
  - ▶ Excessive **Diarrhoea**
  - ▶ **Fistulas**
    - ▶ **Uretoenteric**
    - ▶ **Pancreaticoduodenal**

# Respiratory Acidosis

- ▶ Hypoventilation
- ▶ “Won’t breathe, can’t breathe, can’t breathe enough”
- ▶ “Won’t breathe” – central causes
  - ▶ OD
  - ▶ Traumatic brain injury
  - ▶ CVA
- ▶ “Can’t breathe” – neuromuscular
  - ▶ GBS
  - ▶ Envenomation
- ▶ “Can’t breathe enough” – primary respiratory pathology
  - ▶ Obstruction
  - ▶ COPD
  - ▶ Asthma
  - ▶ Lung insults
    - ▶ Trauma
    - ▶ Lung membrane diseases eg. Pulmonary oedema, LRTI, ARDS, aspiration
- ▶ Acute vs Chronic



# Respiratory alkalosis

- ▶ Hyperventilation
- ▶ Hypoxaemia
  - ▶ Altitude
  - ▶ PE
  - ▶ Pneumonia
  - ▶ Asthma
  - ▶ Pulmonary oedema
- ▶ Central
  - ▶ Voluntary or psychogenic (anxiety, pain)
  - ▶ CVA
  - ▶ Increased ICP (tumour, ICH, trauma)
- ▶ Tox
  - ▶ Salicylates
  - ▶ Caffeine, nicotine
  - ▶ Sympathomimetics
  - ▶ Thyroxine
- ▶ Increased metabolic state eg. pregnancy

# Metabolic Alkalosis

- ▶ H<sup>+</sup> loss
  - ▶ HCl loss
    - ▶ Vomiting
  - ▶ Drugs
    - ▶ Diuretics eg. Frusemide
    - ▶ Laxative abuse
- ▶ Alkali gain
  - ▶ NaHCO<sub>3</sub> therapy
- ▶ Impairment of kidney's ability to excrete HCO<sub>3</sub>
  - ▶ Cl depletion
  - ▶ K depletion
    - ▶ Endocrinopathies
      - ▶ Cushing's
      - ▶ Bartter's
  - ▶ Decreased GFR
  - ▶ ECF volume depletion

# Assess Oxygenation

▶ ABG

▶ Aa gradient

$$= \text{PAO}_2 - \text{PaO}_2$$

$$= \frac{\text{age}}{4} + 4$$

▶ PAO<sub>2</sub>

$$= ((760 - \text{vapour pressure}) \times \text{FiO}_2) - (\text{paCO}_2 \times 1.25)$$

$$= (713 \text{ (sea level)} \times \text{FiO}_2) - (\text{paCO}_2 \times 1.25)$$

$$= 150\text{mmHg (sea level, room air)} - (\text{paCO}_2 \times 1.25)$$

# Check other electrolytes

## ▶ Na

- ▶ Correct sodium for glucose eg DKA

- ▶ Corrected Na =  $\text{Na} + \frac{\text{glucose} - 5}{3}$

## ▶ K

- ▶ Correct potassium for pH
- ▶ For each 0.1 fall in pH K should rise by 0.5mmol above 5.0mmol/L
- ▶ For each 0.1 rise in pH K should fall by 0.5mmol below 5.0mmol/L

## ▶ Cl

- ▶ Glucose

## ▶ Osmolality

- ▶  $2 \times \text{Na} + \text{urea} + \text{glucose}$
- ▶ Osmolar gap = measured osmolality – calculated osmolality

# Interpretation of blood gases

- ▶ Acid base balance
  - ▶ pH, CO<sub>2</sub>, HCO<sub>3</sub>
  - ▶ Major disturbance
  - ▶ Compensation
  - ▶ Metabolic acidosis
    - ▶ Anion gap
    - ▶ Delta ratio
- ▶ Oxygenation
- ▶ Other electrolytes